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8 2 collision theory and reaction mechanisms chemistry

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the first step in the gas phase reaction between carbon monoxide and oxygen is a collision between the two molecules $\text{CO} + \text{O}_2 \rightarrow \text{CO}_2 + \text{O}$ although there are many different possible orientations the two molecules can have relative to each other consider the two presented in figure 8 2 1 8 2 1

2 7 reaction mechanisms elementary reactions

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the sequence of events that occur at the molecular level during a reaction is the mechanism of the reaction it describes how individual atoms ions or molecules interact to form particular products the stepwise changes are collectively called the reaction mechanism each step of the reaction is called an elementary reaction

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