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## separations and purifications questions khan academy

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separations and purifications questions 500 ml of water and 500 ml of dichloromethane d 1 3 g ml are added to a mixture containing benzoic acid cresol methoxyethane and n methylethanamine four solutions are available for extraction of the mixture hcn hcl naoh and lihco 3

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vacuum filtration suction filtration is a chemistry laboratory technique which allows for a greater rate of filtration whereas in normal filtration gravity provides the force which draws the liquid through the filter paper in suction filtration a pressure gradient performs this function

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answer the correct answer is d because the difference in solubility in water means the first step is to make a solution the magnesium chloride will dissolve but the solid calcium sulfate will be left behind the mixture is filtered to remove the calcium sulfate and then evaporated and crystallised to obtain magnesium chloride crystals

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this set contains simple techniques to separate the substances in a mixture learn with flashcards games and more for free

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five common techniques are distillation fractional distillation filtration crystallisation and chromatography simple distillation this is used to separate a liquid and soluble solid from a solution e g water from a solution of salt water or a pure liquid from a mixture of liquids

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separation techniques if a substance does not dissolve in a solvent we say that it is insoluble for example sand does not dissolve in water it is insoluble filtration is a method for separating an insoluble solid from a liquid when a mixture of sand and water is filtered the sand stays behind in the filter paper it becomes the residue

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for liquid liquid mixtures containing multiple different liquids a range of techniques such as fractional distillation can be used depending on the properties of the liquids involved described below are some of the most common techniques used to separate mixtures

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basic techniques calorimetry calorimetry is used to measure the change in heat released or absorbed during a chemical reaction in chemistry labs the setup for calorimetry generally consists of a styrofoam cup with a sealed lid and hole for thermometer to measure the change in temperature during the reaction centrifugation

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google classroom a chemist is separating a mixture of two substances x and y using column chromatography the chemist loads the mixture onto a column filled with silica gel a polar material and then flows a nonpolar solvent through the column

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