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a solution is a homogeneous mixture the major component is the solvent while the minor component is the solute solutions can have any phase for example an alloy is a solid solution solutes are soluble or insoluble meaning they dissolve or do not dissolve in a particular solvent

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when a solution contains the maximum amount of solute that can dissolve under a given set of conditions it is a saturated solution otherwise it is unsaturated supersaturated solutions which contain more dissolved solute than allowed under particular conditions are not stable the addition of a seed crystal a small particle of solute

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course objectives this class will continue coverage of fundamental concepts of organic chemistry including infrared ir and nuclear magnetic resonance nmr spectroscopy the structure nomenclature and reactivity of organometallic compounds aldehydes ketones carboxylic acids and their derivatives enolates and enamines aromatic compou



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