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complete list of inorganic acids formula acid name salt name formula acid name salt name h₂so₃
sulfurous acid

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common strong acids include hydrochloric acid sulfuric acid phosphoric acid and nitric acid common weak acids include acetic acid boric acid hydrofluoric acid oxalic acid citric acid and carbonic acid here is a list of ten common acids with their chemical structures

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acids are molecular compounds that release hydrogen ions a binary acid consists of hydrogen and one other element oxyacids contain hydrogen oxygen and one other element the name of the acid is based on the anion attached to the hydrogen

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since all acids contain hydrogen the name of an acid is based on the anion that goes with it these anions can either be monatomic or polyatomic the name of all monatomic ions ends in ide

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it should not be hard for you to name several common acids but you might find that listing bases is a little more difficult below is a partial list of some common acids along with some chemical formulas

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what is an acid in chemistry definition and examples this entry was posted on february 14 2023 by anne helmenstine updated on january 13 2024 there are different types of acids by definition an acid donates hydrogen ions or protons or accepts an electron pair

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examples of acids include the inorganic substances known as the mineral acids sulfuric nitric hydrochloric and phosphoric acids and the organic compounds belonging to the carboxylic acid sulfonic acid and phenol groups

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as a binary acid the acid s name is hydro stem name ic acid because this acid contains a bromine atom the name is hydrobromic acid because this acid is derived from the sulfate ion the name of the acid is

the stem of the anion name ic acid the name of this acid is sulfuric acid test yourself name each acid hf hno₂ answers

list of common strong and weak acids thoughtco Jul 10 2023

thoughtco by anne marie helmenstine ph d updated on may 06 2024 strong and weak acids are important to know both for chemistry class and for use in the lab there are very few strong acids which are acids that completely dissociate or ionize in an aqueous solution or water

formulas of common acids and bases thoughtco Jun 09 2023

updated on july 03 2019 acids and bases are used in many chemical reactions they are responsible for most color change reaction and are used to adjust the ph of chemical solutions here are the names of some of the common acids and bases and the formulas associated with them formulas of binary acids a binary compound consists of two elements

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acids are very common in some of the foods that we eat citrus fruits such as oranges and lemons contain citric acid and ascorbic acid which is better known as vitamin c carbonated sodas contain

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naming acids since all acids contain hydrogen the name of an acid is based on the anion that goes with it these anions can either be monatomic or polyatomic the name of all monatomic ions ends in ide the majority of polyatomic ions end in either ate or ite though there are a few exceptions such as the cyanide ion cn

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acid name formula acetic acid ch₃cooh antimonic acid h₃bo₃ antimonous acid h₃sbo₃ arsenic acid h₃aso₄ boric acid h₃bo₃ bromic acid hbro₃ bromous acid hbro₂ carbonic acid h₂co₃ carbonous acid h₂co₂ chloric acid hclo₃ chlorous acid hclo₂ chromic acid h₂cro₄ chromous acid h₂cro₃ citric acid c₆h₈o₇ cyanic acid hcn

6 5 naming acids enhanced introductory college chemistry Feb 05 2023

6 5 naming acids enhanced introductory college chemistry learning objectives by the end of this section you will be able to define acid name a binary acid and an oxyacid there is one other group of compounds that is important to us acids and these compounds have interesting chemical properties

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molar calculations percent composition and empirical formula the name of acid is derived from the name of its anion each ending and prefix of an anion forms a specific combination in the name of the acid

5 9 naming acids chemistry libretexts Dec 03 2022

formula for naming acids hydro and base name of nonmetal and ic acid example hcl is hydrochloric acid naming oxyacids an oxyacid is an acid that consists of hydrogen oxygen and a third element the third element is usually a nonmetal a oxyanions with ite ending

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acid definition and examples in chemistry thoughtco Oct 01 2022

updated on january 13 2020 an acid is a chemical species that donates protons or hydrogen ions and or accepts electrons most acids contain a hydrogen atom bonded that can release dissociate to yield a cation and an anion in water

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acids and bases acids and bases acids and bases are popular chemicals which interact with each other resulting in the formation of salt and water the word acid comes from a latin word acere which means sour table of contents acids and bases definition recommended videos theories of acids and bases ph of acids and bases

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